ADVANCED CATALYTIC STUDIES FOR POWER ENGINEERING AND ENVIRONMENTAL PROTECTION

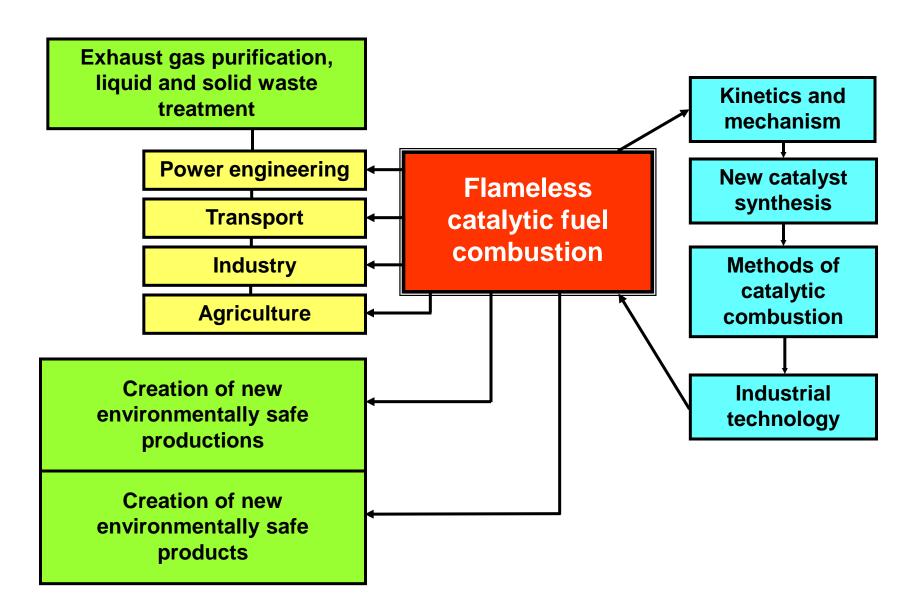
Z.R. Ismagilov

Boreskov Institute of Catalysis

CONTENTS OF THE LECTURE

- Introduction
- Catalytic combustion
- Catalytic heaters
- Catalytic boilers
- Catalytic fluidized bed combustion
- Catalytic two-stage combustion
- Catalytic combustion for Gas Turbines
- Fuel Cells

CATALYSIS FOR POWER ENGINEERING AND ENVIRONMENT



KEY ROLE OF CATALYSIS

Fuel production - 100%

Fuel combustion - >0,1%

Fuel cells - 100%

Exhaust abatement - 80%

CATALYSIS FOR POWER ENGENEERING

Increase combustion efficiency

Improve environmental protection

FUEL COMBUSTION

Traditional Combustion (Chain Mechanism)

RH + O₂
$$\xrightarrow{\text{flame}}$$
 CO₂ + H₂O + Q₁ \longrightarrow CO + Benz[α]pyren

N₂ + O₂ \longrightarrow NO, NO₂ \longleftarrow thermal NO_x

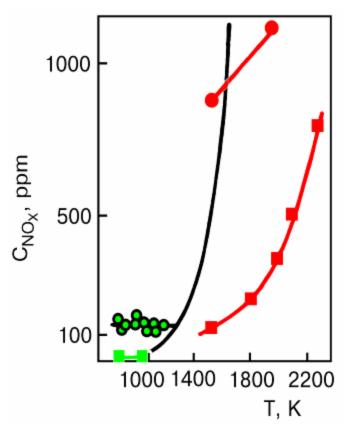
RNH + O₂ \longrightarrow NO x + ... \longleftarrow fuel NO_x

Heterogeneous Catalytic Oxidation (Low Temperature Flameless Combustion)

RH + O₂
$$\xrightarrow{\text{catalyst}}$$
 CO₂ + H₂O + Q₂
N₂ + O₂ $\xrightarrow{\text{kono}}$ NO_X

RNH + O₂ $\xrightarrow{\text{kono}}$ N₂ + CO₂ + H₂O

RNH + O₂ $\xrightarrow{\text{kono}}$ NO_X + CO₂ + H₂O



Dependence of NO_x concentration on temperature at combustion of fuel containing 1 wt.% of bound nitrogen:

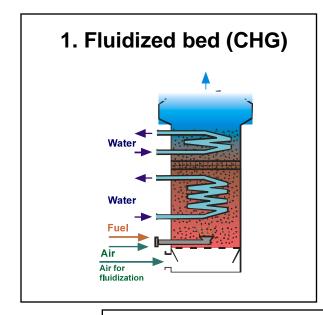
Advantageous Fields of Catalytic Combustion Development

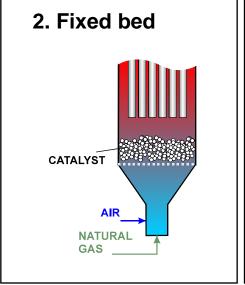
- Greenhouse effect
 - energy saving industrial technologies based on catalytic combustion, to diminish CO₂ formation.
- Environmentally safe transport
 - new generation of engines on the principles of catalytic fuel combustion;
 - catalytic purification of automotive exhaust: combustion of hydrocarbons, CO, aldehydes, soot particles, etc.
- Environmental problems of energy production
 - catalytic combustion in gas turbines and boilers;
 - catalytic fluidized bed combustion;
 - two-stage catalytic combustion;
 - catalytic space heaters, household appliances.
- □ VOC Control
- □ Environmental impact of oil and gas mining, transportation

MAIN APPLICATIONS OF CATALYTIC FUEL COMBUSTION

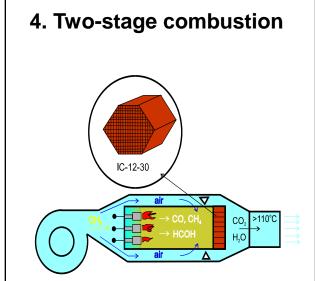
- Catalytic gas heaters
- Boilers
- Two-stage combustion
- Gas turbines
- Fuel cells

TYPES OF CATALYTIC BURNERS











CATALYTIC GAS HEATERS

Catalyst support

- Fiber materials
- Ceramic monoliths

Active component

- Low ignition temperature
- High thermal stability

Catalytic Gas Heaters «Termokat»



CATALYTIC BURNER



Mn containing catalyst on highly porous foam material, diameter 300 mm

CATALYTIC COMBUSTION GAS BOILER

Distinctive features:

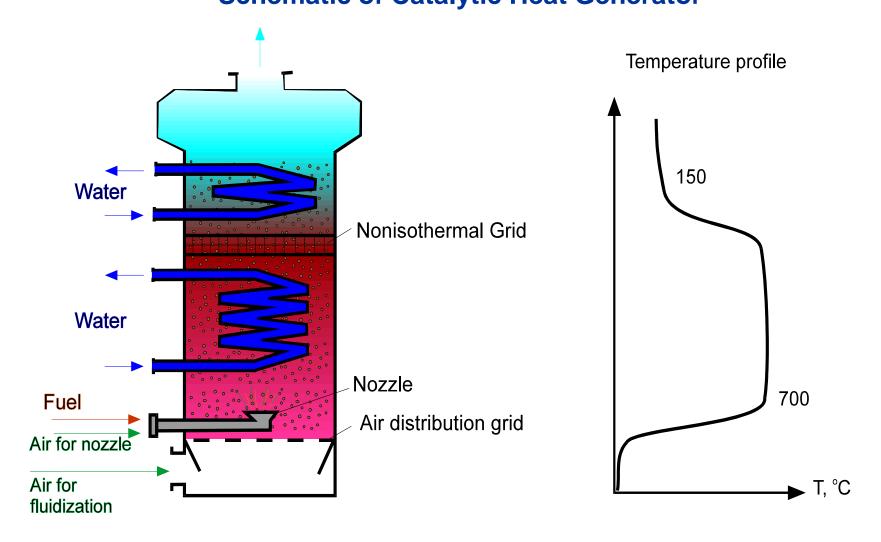
- Full combustion of natural gas;
- No nitrogen oxides in the exhaust;
- Carbon oxide concentration is below 0,01 vol. %.

Technical characteristics:

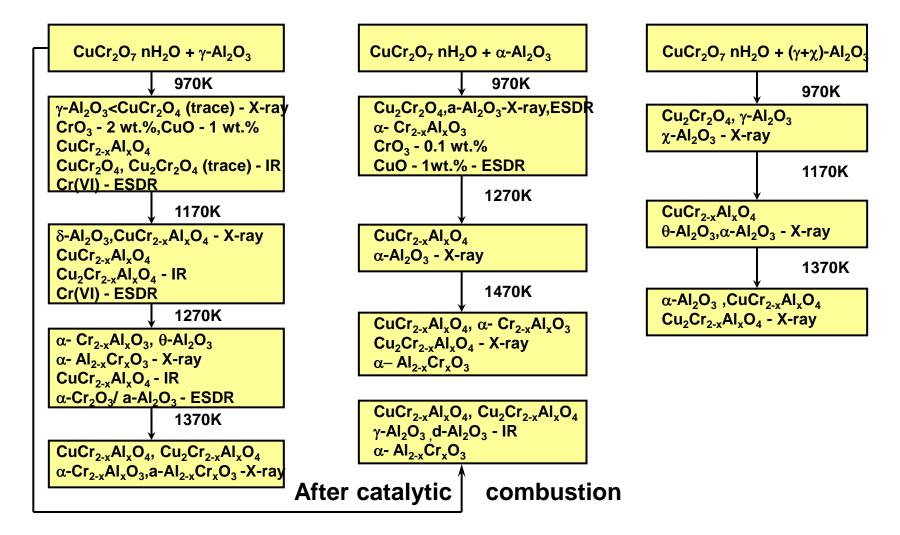
Rated heat power, kW	16
Efficiency, %	88-94
Water temperature at the output, °C	50-90
Flue gas temperature, °C, not less	110
Natural gas consumption at	
1270 Pa pressure, m ³ /hour	1.6
Dimensions, mm 890x44	0x440
Mass, kg	90

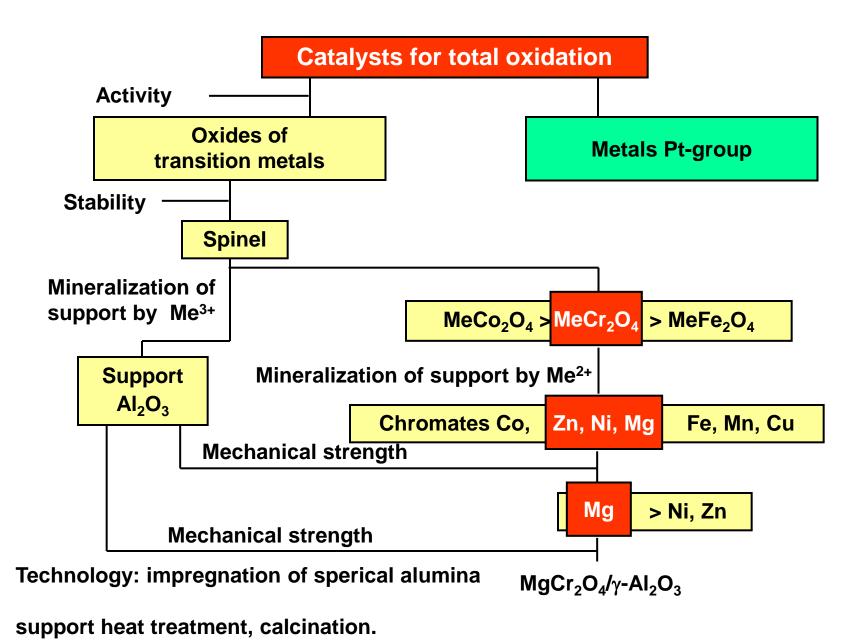


Fluidized Bed Catalytic Combustion Schematic of Catalytic Heat Generator



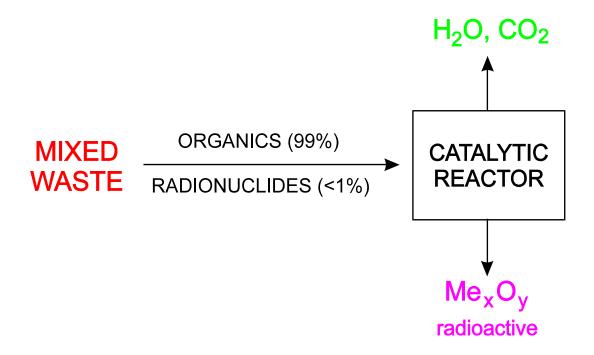
Scheme of Phase Transitions in Supported Copper-Chromium Catalyst





Catalytic Destruction of Mixed Organic Radioactive Wastes

- complete destruction of hazardous organic components without secondary emissions
- compacting, more than 10000 fold reduction of volume of radioactive waste for further processing by existing technologies, vitrification



PROTOTYPE DEMONSTRATION PLANT FOR MIXED ORGANIC WASTE TREATMENT AT THE PLANT OF CHEMICAL CONCENTRATES, NOVOSIBIRSK



Catalytic reactor

Catalytic fluidized bed destruction of toxic rocket fuel

Objective

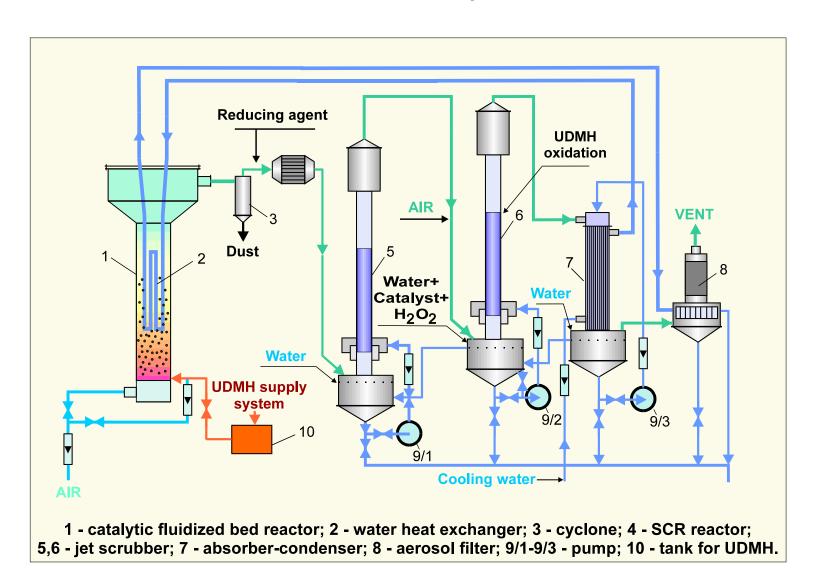
Environmentally safe and efficient utilization of an extremely toxic and explosive rocket fuel 1,1-dimethylhydrazine:

unsymmetrical dimethylhydrazine, UDMH, technical name - heptyl

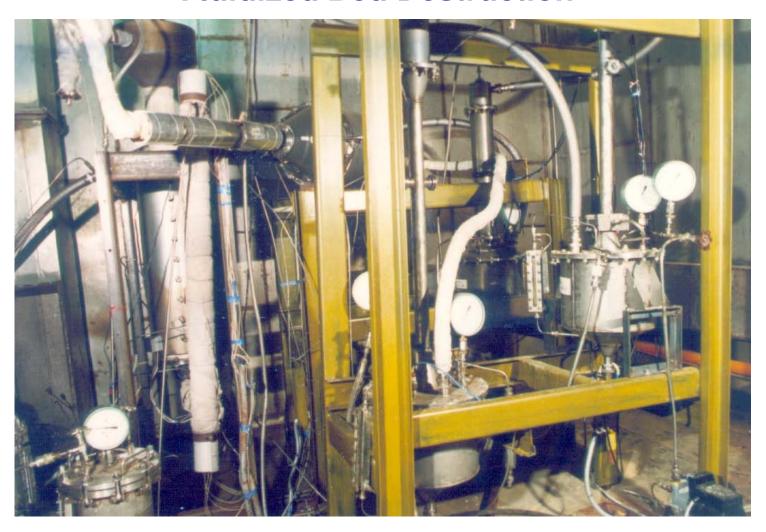
Method

Total catalytic oxidation of UDMH by air in a fluidized bed reactor

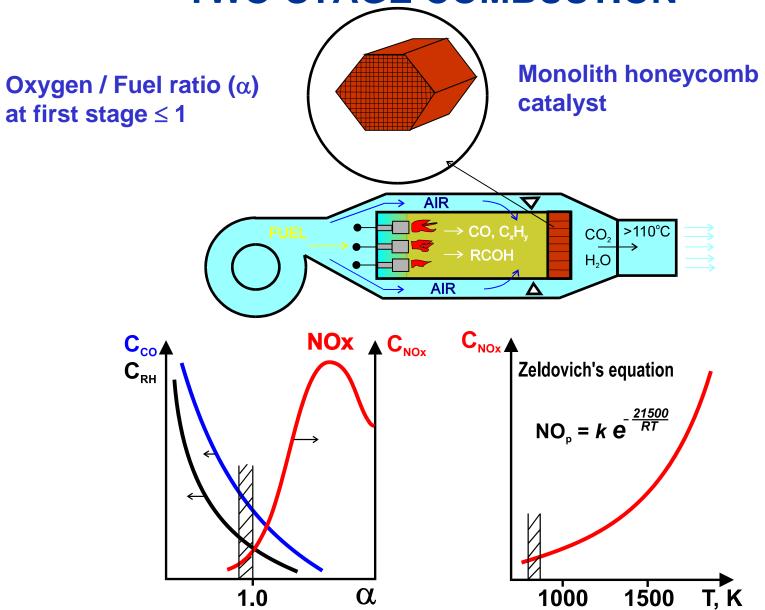
Schematics of a Pilot Plant for Catalytic Utilization of UDMH



Photograph of the Pilot Plant for UDMH Catalytic Fluidized Bed Destruction



CATALYTIC AIR HEATER TWO-STAGE COMBUSTION



TWO STAGE COMBUSTION

Catalytic cartridges

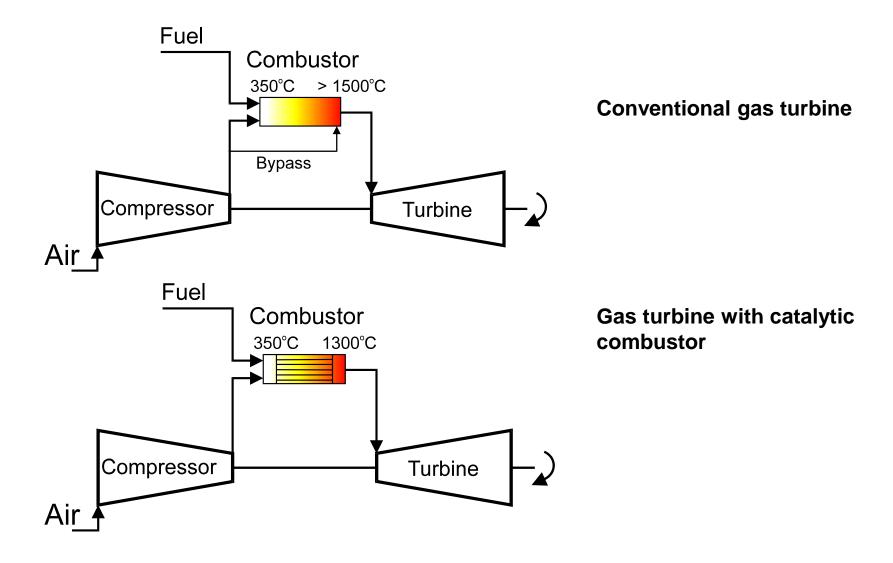


Two-stage catalytic heat generators has been successfully used for greenhouse heating at farm "Priobskoe" near Novosibirsk, Russia during the last three years

Catalytic gas turbines is a new approach in environmentally clean power production

The use of a catalytic burner in a gas turbine allows:

- reduction of combustion temperature
- stabilization of combustion of lean mixtures
- reduction of NO_x, CO and HC emissions



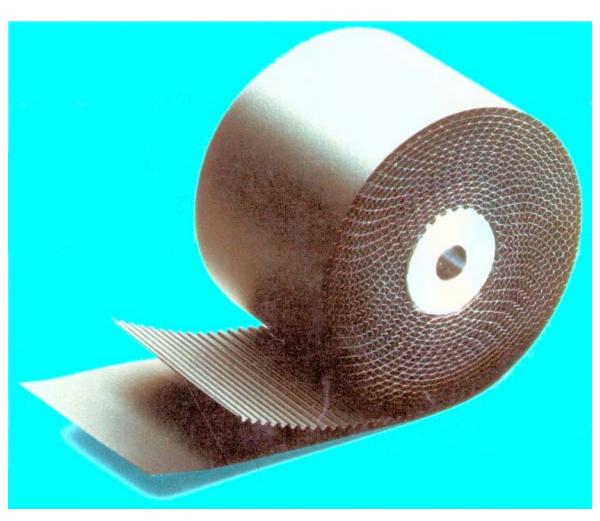
Catalytica combustion Systems, Inc.

www.energysolutionscenter.org/DistGen/AppGuide/DataFiles/Xonon.pdf

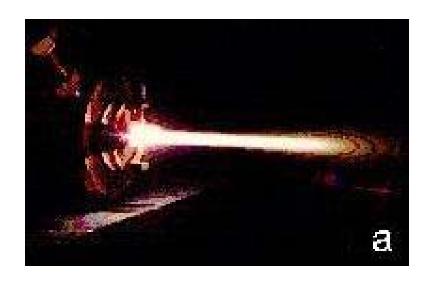
Requirements to catalysts for gas turbines

- high thermal stability (900-1000°C)
- resistance to thermal shocks
- high mechanical strength P=10 atm, velocity 40 m/s
- pressure drop less than 3%.

METAL MONOLITHIC SUPPORT



Plasma spray coating of catalyst on metal foil





Laminar (a) and turbulent (b) plasma jet outflow.

Micrograph of cross section view of Al₂O₃ plasma sprayed on metal surface

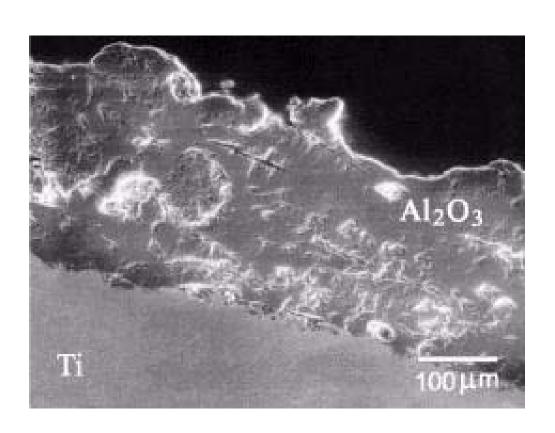


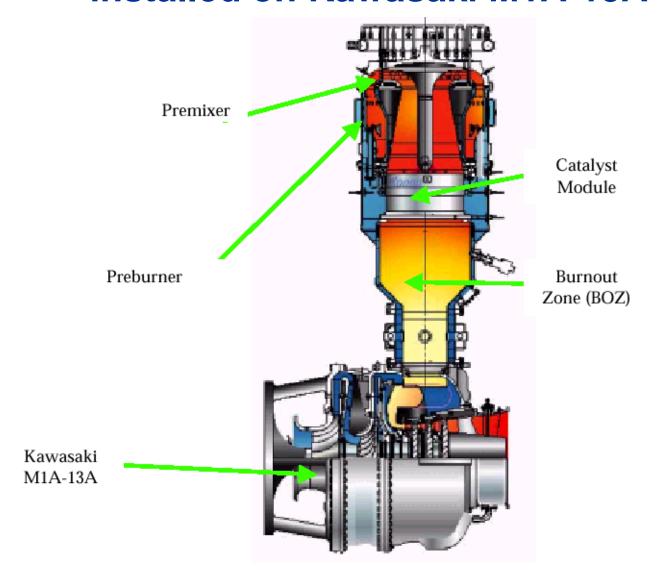
Photo of metal samples with plasma sprayed Al₂O₃



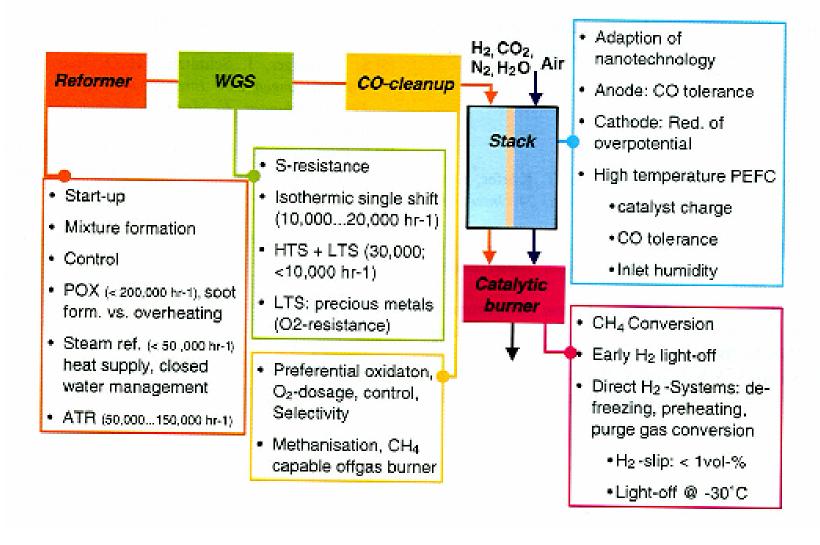
METAL MONOLITH CATALYTIC BURNER



Xonon® Catalytic Combustion System installed on Kawasaki M1A-13A



Catalyst application in fuel cell systems: development focus



Main areas of catalysis application for fuel cell technology

- Hydrogen production from natural gas and other fuels
- Catalysis of electrochemical reactions in fuel cell electrodes

Methods of syn-gas production from methane

1. Steam methane reforming

$$CH_4 + H_2O = CO + 3 H_2 - 206 kJ/mole$$

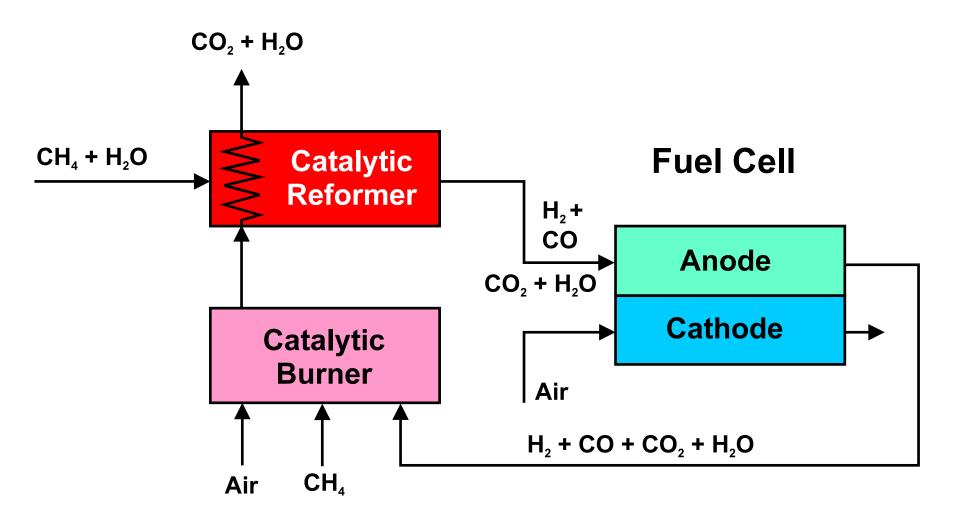
2. Dry methane reforming

$$CH_4 + CO_2 = 2 CO + 2 H_2 - 247 kJ/mole$$

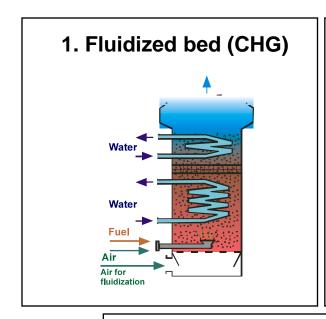
3. Catalytic partial oxidation

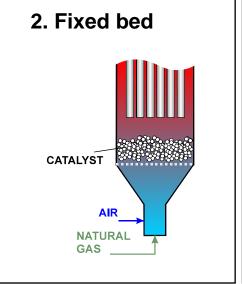
$$CH_4 + 1/2 O_2 = CO + 2 H_2 + 36 kJ/mole$$

Schematic diagram of the process

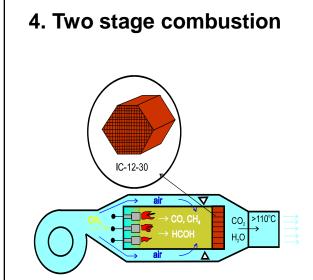


TYPES OF CATALYTIC BURNERS



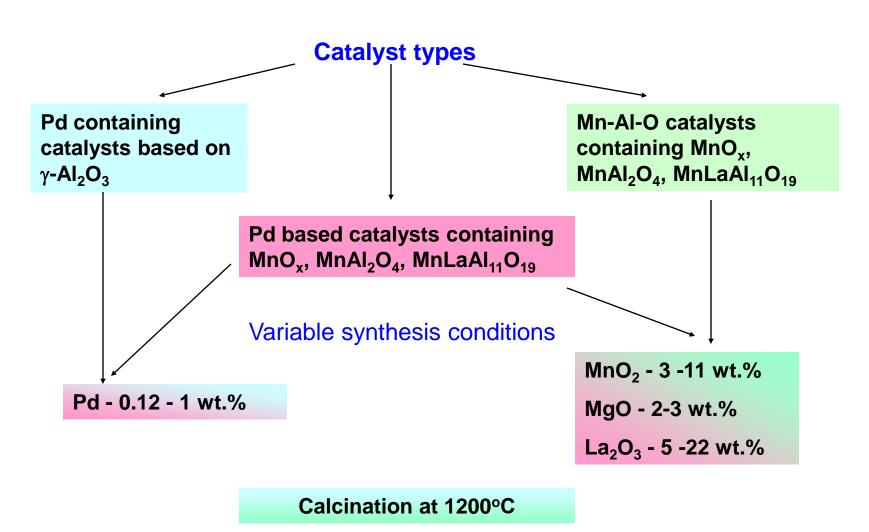




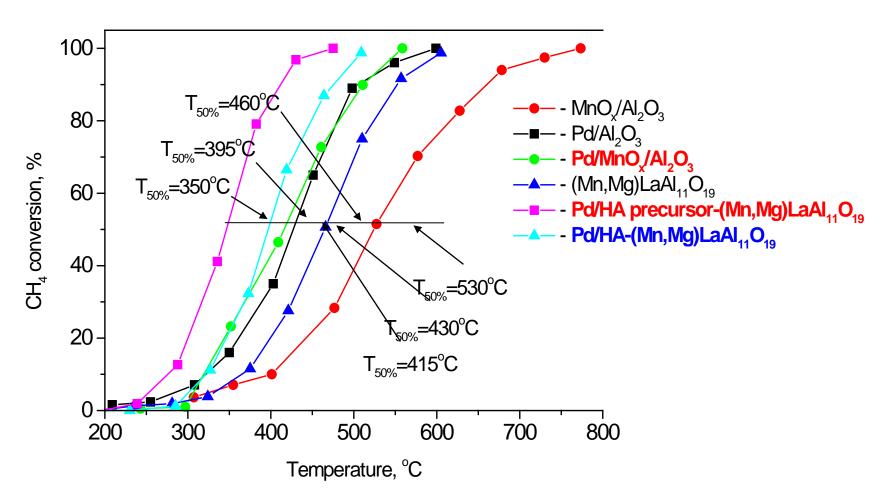




Synthesis of catalysts for combustion of HC fuels. Systematic variation of synthesis conditions



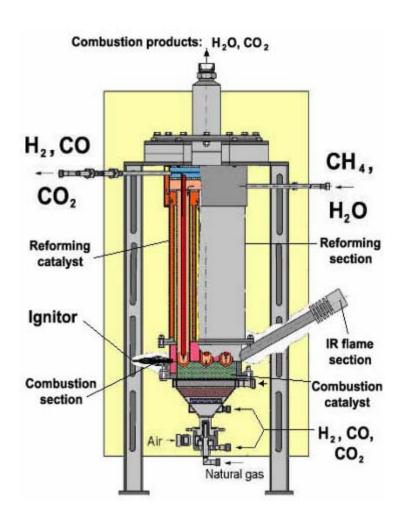
Synergetic effect of Pd and oxide manganese compounds: MnO_x, MnAl₂O₄, MnLaAl₁₁O₁₉

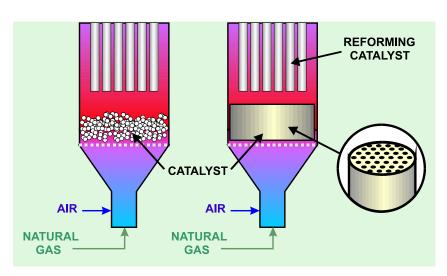


 $Pd/MnO_x/\gamma-Al_2O_3$ ($T_{50\%}=415$ °C) ~ $Pd/\gamma-Al_2O_3$ ($T_{50\%}=430$ °C) <

 $Pd/HA-(Mg,Mn)LaAI_{11}O_{19} (T_{50\%}=395^{\circ}C) < Pd/HA \ precursor-(Mn,Mg)LaAI_{11}O_{19} (T_{50\%}=350^{\circ}C) < Pd/HA \ pr$

Catalytic burners for reformers of fuel cell energy installations





Two types of catalysts